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L4: Entry 37 of 37

File: USPT

Oct 30, 1973

US-PAT-NO: 3769175

DOCUMENT-IDENTIFIER: US 3769175 A

TITLE: PROCESS AND APPARATUS FOR THE CONTINUOUS TREATMENT OF LIQUIDS WITH ENZYME CARRIERS

DATE-ISSUED: October 30, 1973

INVENTOR-INFORMATION:

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PARENT-CASE:

This is a divisional application of Ser. No. 758,019 filed Sept. 6, 1968, and now abandoned.

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REF-CITED:

U.S. PATENT DOCUMENTS

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	PAT-NO	ISSUE-DATE	PATENTEE-NAME	US-CL
<input type="checkbox"/>	<u>3580840</u>	May 1971	Uridil	210/11
<input type="checkbox"/>	<u>3425839</u>	February 1969	Pinnegar	99/31
<input type="checkbox"/>	<u>2811336</u>	October 1957	Bready	257/234

ART-UNIT: 172

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ASSISTANT-EXAMINER: Penland; R. B.

ATTY-AGENT-FIRM: Curtis, Morris & Safford

ABSTRACT:

A process and apparatus has been provided for accelerated fermentation of fermentable liquids, such as beer wort. Accordingly, an enzyme carrier dispersion such as yeast is deposited either before or concurrently with the fermented liquid in a layer on a porous body which has pores sized to pass the liquid but not the enzyme carrier. As a result of the capability to support considerable amount of the enzyme carrier on the porous support, fast fermentation rates are achieved. Moreover, the enzyme carrier may be deposited on an inert material and then deposited on the porous support which allows easy metering of the necessary amount of enzyme for facile control of the fermentation reaction. An apparatus has also been provided which comprises at least one chamber having an inlet conduit for introducing an enzyme carrier dispersion and the liquid, the chamber being provided with communicating conduits with a next, adjacent chamber. Further, closing means are provided for the apparatus communicating with each chamber for alternatively closing said inlet conduits, as well as a porous member within the chamber upon which the enzyme carrier is deposited. The porous member has pores of a diameter less than the diameter of particle sizes in said enzyme dispersion. Still further, the apparatus is provided with a conduit for collecting the fermented liquid communicating with the last chamber, the porous body having communicating conduits with each chamber wherein the porous body forms the outlet opening of the preceding chamber and the inlet opening for the adjacent chamber.

6 Claims, 1 Drawing figures Number of Drawing Sheets: 1

BRIEF SUMMARY:

This invention relates to a process and apparatus for the continuous treatment of liquids having enzyme carriers in admixture therewith; more particularly, this invention pertains to a new process for the continuous treatment of liquids with enzyme carriers in high concentrations, preferably for the fermentation of alcoholic beverages such as wine, beer, champagne or for the fermentative ripening of cream for the production of cheese by means of rennet as well as to an apparatus for the carrying out of this method.

A continuous fermentation of wort in the presence of yeast in the manufacture of beer is known from British Pat. No. 872,395 and from German Provisional Pat. Nos. 1,205,041 and 1,207,324. The first two patents relate to two-stage fermentation processes in which a mixture of wort, yeast and air is subjected to fermentation in a first tank while intensively and continuously agitated, and the mixture is fermented further in a second tank with weaker agitation, whereupon the yeast is separated and the fermented wort is discharged after cooling and separation of the yeast. The last-mentioned German Provisional Patent describes a continuous fermenting of wort in the presence of yeast of high concentration in a slim fermentation tower. This tower is provided at its lower end with inlets for wort and air and is connected at its upper end to a yeast settling chamber, perforated distribution elements, which extend over the cross-section of the fermentation tower, being arranged over the entire length of the latter.

With such an apparatus, it is said to be possible to operate with yeast concentrations of about 20 to 60 percent and, hence, correspondingly accelerate the fermentation process.

However, these methods have the disadvantage that yeast, which acts as enzyme carrier, is suspended in the wort so that the enzyme is present in the wort to be fermented only in a relatively low concentration of about 2 to 5 percent. As it is known, however, that the rate of fermentation is a function of the yeast, i.e., enzyme concentration, these processes do not offer any advantage in this respect over batch methods.

In accordance with German Provisional Pat. No. 1,207,324, the process apparently is carried out with very high yeast concentrations of between 20 and 60 percent while attempting to maintain the yeast in the fermentation tower in a suspended condition by supporting it directly by means of supporting elements and by distributing the flow of the fermentation gases over the cross-section of the

tower. Even in this method, the enzyme concentration is high only at the start of the process while as the process is continuously carried out, the yeast very rapidly is carried along upwards into the settling chamber and is thereby diluted.

Another disadvantage of these processes is that the fermented fluid, i.e., beer, is insufficiently separated from the yeast so that an additional filtration step is required.

It is the object of this invention to provide a new process for the continuous treatment of liquids having in admixture therewith enzyme carriers, e.g., for the continuous treatment of wort with yeast, in which the fermentation times are very substantially reduced by increasing as much as possible the enzyme concentration in the fermentation chamber or zone. Another object of the novel process is to provide an apparatus which takes up as little space as possible, which can easily be handled, and from which, after the consumption of the enzyme carrier, the latter can be removed completely without any complex, tedious processing. Finally, in accordance with the novel process, it is also possible to operate at higher temperatures, i.e., above the previously known optimum conditions for the enzyme carrier in question and with increase in pressure, thus, further reducing the time of treatment. Finally, another object of the present invention is to provide an especially designed apparatus for the carrying out of this method.

In accordance with the invention, the method for the continuous treatment of a liquid with an enzyme carrier of high concentration consists in conducting the liquid through a deposit of an enzyme carrier which is on at least one porous body and then conducting the liquid through said porous body, the pores of which are so dimensioned or sized that the enzyme carrier is retained substantially completely on the porous body but permits passage of the liquid to be treated.

When the expression "enzyme carrier" is used herein, it is to be understood thereby preferably any enzyme-producing microorganism such as yeast, as well as synthetic enzyme carriers such as for instance adsorbent carrier materials on which the enzyme has been adsorbed by immersion of the carrier materials in the enzyme solution.

For the sake of simplicity, the method will be explained below on basis of the fermentation of wort in presence of yeast to form beer. However, it is expressly pointed out that the method is also suitable for any other treatment of liquids with enzymes such as: additionally, the fermentation of other alcoholic beverages such as wine or champagne; or the acid ripening of cream for the production of cheese by means of rennet.

In the process of the present invention, it is preferable to proceed by pumping an enzyme carrier, e.g., yeast, in an aqueous suspension through an inlet opening of a fermentation chamber, the outlet opening of which consists of a porous body, such as a frit-like porous plate. As the pores of the porous body are so small for Example 3 m.mu. that the enzyme carrier is retained on it in the fermentation chamber while the carrier-suspending liquid flows through it, the enzyme carrier collects as a deposit on the porous body. Thereupon, the liquid to be treated, in this case wort for the obtaining of beer, is conducted continuously through the fermentation chamber until the enzyme carrier in the fermentation chamber is used up. Thereupon, the flow of the treated liquid is interrupted for a short time and the spent enzyme carrier is removed from the fermentation chamber, for example by back-washing. Thereafter, suspended enzyme carrier is again forced into the fermentation chamber and the liquid to be treated such as wort is again continuously passed through the same. A continuous separation of spent enzyme carrier is, however, also possible in a rotary filter apparatus.

It is evident that the temporary interruption instead of the back-washing step of the novel, continuous process, in order to remove the spent enzyme carrier, lasts only for a short time and, therefore, scarcely detracts from the continuous operation which is carried out in actual practice.

Instead of introducing the enzyme carrier into the fermentation chamber before passage of the liquid which is to be treated, the enzyme carrier can also be added to the liquid itself in the quantity necessary for the depositing on the porous body, in which case it also deposits on the porous body at the outlet end of the fermentation chamber.

It is particularly advisable in the process of the invention to pass the liquid to be treated through two or more fermentation chambers connected in series. The speed of fermentation is thereby greatly increased. Moreover, the destruction of the enzyme, which occurs constantly during the fermentation resulting in the reduction of the enzyme concentration in the fermentation chamber, can be overlooked during a large part of the treatment period. Operation with a plurality of porous bodies connected in series with each other, and having enzyme carriers deposited on them, or preferably with a plurality of fermentation chambers connected in series, has the further advantage that it is possible to operate with different enzyme carriers, such as different types of yeast, which lead to a different rate of fermentation.

In the present process, the liquid is preferably forced under pressure through the treatment chambers; for Example at pressure from about 0.5 to 5 atm. Furthermore, it is advantageous to operate at elevated temperatures because in this manner the speed of fermentation can further be increased.

In the case of previously known fermentation processes, it was believed that the temperature of the liquid to be treated, such as wort for the production of beer, cannot be increased beyond the optimum temperature known for the enzyme carrier used without the enzyme carrier being destroyed to a considerably extent and the rate of fermentation, thus, reduced. In the present method in which the enzyme concentration in the treatment zone is very high, it was found that a destruction of a part of the enzyme carrier caused by increasing the temperature above the previously believed optimum for the enzyme carrier in question for Example 50 instead 30.degree. Celsius does not bring about any destruction of the enzyme (which has a higher optimum temperature than the enzyme carrier) nor bring about any detrimental consequences to the process; and the advantages of the accelerating of the fermentation effected by the increase in temperature far outweigh the disadvantages of the destruction of a part of the enzyme carrier.

Accordingly, it is particularly advantageous in the novel process to increase the temperature of the treated liquid, for instance by heating elements arranged in the fermentation chambers, to a temperature which is above the previously known optimum temperature for the enzyme carrier in question. Furthermore, when using a plurality of treating chambers, temperatures can be used which vary from chamber to chamber.

Finally, in many cases it is also advisable to treat the liquid, such as wort for the obtaining of beer, by blowing oxygen or nitrogen into the fermentation chambers whereby the speed of fermentation can further be increased.

The advantages of the present method over the previous methods reside primarily in the fact that the times of residence in the treatment zone can be very greatly reduced. This is possible as a result of the high enzyme concentration in the enzyme carrier deposit in the treatment chamber since the chemical reactions such as the fermentation or acid ripening of cream take place very much faster thereby.

For this reason, apparatuses which are very much smaller are required when compared to the prior art apparatuses for the obtaining of equal quantities of treated liquid. Moreover, operating time and labor are also saved. By the rapid passage of the liquid through the treatment zone, increase in the process pressure, and working at temperatures beyond the previously known optimum, a further reduction of the treatment time or residence time is obtained, thus, counteracting the disadvantages resulting from an increased consumption of enzyme carrier and thereby being able to neglect the same. Another advantage of the novel method is that with the small size of an apparatus required the replacement of the enzyme carrier after a certain period of time (which is always necessary even in known methods) can be carried out easily and fast.

In accordance with the invention another preferred embodiment of the process consists in using an enzyme carrier which is adsorbed on a surface-active, inert carrier material and, thus, is of a fixed quantity. This facilitates the carrying out of the fermentation since the enzyme carrier, such as yeast, can be accurately dosed, i.e., measured.

Such an enzyme carrier which can be dosed in a given quantity is obtained by

mixing the enzyme-supplying micro-organisms, such as yeast, in aqueous suspension with a surface-active, inert carrier material and after adsorption on the carrier material subjecting same to a careful drying.

Surprisingly, it has, thus, been found that a definition by weight of the number of bacteria in the microorganisms is possible when these are adsorbed on surface-active material such as silica, silica-gel, activated carbon, asbestos, kieselguhr, or perlite. These surface-active materials can be so developed with respect to the size of their surface or so selected by mesh sizes that a given quantity of the said materials adsorbs a given determinable number of microorganisms. The determination of this value can be effected quantitatively by empirical methods in the laboratory and provides reproducible values.

For example, suitable materials for this purpose are pyrogenic silicas (highly dispersed, very pure silicas), having a surface of 175 ± 25 m²/g (by the BET method) of a particle size of between 10 and 40 μ ; or pyrogenic silicas having a surface of 300 m²/g with a size of the primary particles of 5 to 20 μ , 380 \pm 40 m²/g with a size of the primary particles of 3 to 15 μ , or 460 \pm 50 m²/g; or else oxide mixtures having a surface (determined by the BET method) of 200 ± 25 m²/g and size of the primary particles between 10 and 40 or 20 to 50 μ . Similarly, for purposes of the novel process suitable materials are pyrogenically obtained, mixed oxides having an SiO₂ content of more than 98.3 percent a size of the primary particles of 20 to 40 or 10 to 30 μ and a surface as determined by the BET method of 60 ± 15 , 80 ± 15 , or 170 ± 30 m²/g.

In the case of pure silicas, BET surfaces of 120 m²/g (diameter of primary particles 28 μ) have proven suitable and BET surfaces of 240 m²/g (primary particle diameter of 16 μ) have proven advantageous. Similar suitability is also demonstrated in the case of calcium silicates (SiO₂ = 47 to 49.9 percent, Al₂O₃ 0.4 to 0.5 percent and a BET surface of 130 m²/g with an average particle size of 35 μ). In the case of active precipitated aluminum silicates, good results have been obtained with a BET surface of 130 m²/g and a primary particle diameter of 30 μ .

By mixing the microorganism suspension with the adsorption material and after careful removal of the moisture, preferably by adsorption drying without heating, a dry mass is obtained which prevents further reproduction in view of the absence of a liquid nutrient and can, thus, be stored without multiplying or at least having a reduced capability for multiplying. This dry mass can be stored without the maintaining of special temperatures and can be dosed out in precise weights whereby a uniform addition of the microorganisms is obtained.

DRAWING DESCRIPTION:

With reference to the accompanying drawing and by way of further explanation of the method of the invention, an apparatus is depicted in the FIGURE thereof which illustrates the advantages of the carrying out of the novel process as well as the use of the apparatus.

DETAILED DESCRIPTION:

In the apparatus, the two pressure covers 1 and 9 are pressed together with the aid of a manually operated pressure device 10. These pressure covers hold between them three fluid treating chambers 6, 6a, and 6b. These treating chambers are connected together in series and are separated from each other by porous plates 5, 5a, and 5b, which are reinforced by supporting elements 4, 4a, and 4b. the conduit 2 which has lateral (side) openings to each of the treating chambers 6, 6a, and 6b serves as a feed line for suspending the enzyme carrier, e.g., in water. By means of a pipe socket 8 or an enveloping pipe which is supported displaceably in the conduit 2 and which also has lateral openings, the inlet openings to the treating chambers can be closed or opened as desired. A second pipe conduit 3 which serves to feed the liquid to be treated into the first treating chamber 6 can be closed by a similar pipe socket 7. The collecting line 11 or conduit serves for the discharge of the completely treated liquid emerging from the last treating chamber 6b.

Upon the operating of the apparatus described, the inlet opening for the liquid to

be treated in the conduit 3 is first of all closed by means of the pipe socket 7. Thereupon, through the pipe conduit 2 and the inlet openings of said conduit to the treating chamber 6, 6a, and 6b, an aqueous suspension of the enzyme carrier is pumped into the treating chambers, the liquid of the suspension being withdrawn through the porous plates 5, 5a, and 5b, and through the collecting conduit 11 and the enzyme carrier collected as a deposit in the treating chambers on the porous plates. Thereupon by means of the pipe socket 8, the inlet opening of the pipe conduit 2 to the treating chambers 6, 6a, and 6b are closed and the inlet opening of the pipe conduit 3 to the first treating chamber 6 is opened by actuation of the pipe socket 7. Then the liquid which is to be treated is pumped through the pipe conduit 3 through the in series connected treating chambers 6, 6a, and 6b, and this liquid emerges from the treating chamber 6b into the collecting conduit 11. In this connection, the liquid in each of the treatment chambers first flows through the deposit of the enzyme carrier on the porous plate and then through the porous plate itself. In the treating chambers 6, 6a, and 6b, moreover, heat exchange plates can be provided (not shown) which have openings through which the liquid flows. These plates can serve either for the cooling or heating of the liquid to be treated and can have a different temperature in each of the treatment chambers.

The features of the apparatus shown in the drawing can be modified in various manners. Thus, for instance, the pipe sockets 7 and 8 shown can be replaced by other shutoff members such as ordinary valves and the treating chambers 6, 6a, and 6b can be provided with separate feed lines for the enzyme carrier so that different enzyme carriers can be introduced into the individual treating chambers.

Thus, the above-described apparatus has at least one chamber with two inlet openings to be closed in succession by closure members, namely one for the enzyme carrier and one for the liquid to be treated, as well as an outlet opening formed by a porous body for the liquid and a collecting line connected with the outlet opening for the liquid emerging from the chamber.

According to the invention, the apparatus has two or more chambers connected in series, each having an inlet opening for the enzyme carrier and an inlet opening and an outlet opening for the liquid to be treated, the porous body which forms the outlet opening of the preceding chamber forming in each case the inlet opening for the following chamber, as well as a collecting line in communication with the outlet opening of the last chamber. If the porous bodies, in accordance with one suitable embodiment of the invention, are provided on the downstream side with supporting elements, such as elements 4, 4a, or 4b, the latter may consist for instance of perforated plates through which the liquid passes directly into the next chamber or these may be solid plates, in which case the liquid is then deflected from these plates into a channel arranged on the side of the chamber and from there into the next chamber.

In this connection, devices for changing the temperature of the liquid to be treated, for instance, for the heating or cooling thereof, can preferably be provided in some or all of the in series-connected chambers. Suitable heating devices can in this connection consist of heat exchange plates arranged parallel to the porous bodies and having openings through which the liquid to be treated passes.

The following example serves further to explain the method of the invention and the advantages obtained thereby.

EXAMPLE

In the laboratory test the above-described apparatus was used with only one chamber for the fermentation of beer wort by means of yeast. At the start of the experiment, washed and pressed brewery yeast was formed into a suspension as a viscous mass in water, and the same introduced into the treating chamber in an amount that a yeast cake of a thickness of 30 mm deposited on the porous plate having a surface of 20 x 20 mm. Thereupon, original beer wort of 13.2 percent total extract, measured by hydrometer (with due consideration of the temperature correction) was pumped through the treating chamber, the yeast cake, and the porous plate. At the start of the experiment, water still contained in the yeast cake was washed out by the beer wort so that the first runnings of the process were discarded. The temperature in the treatment chamber was maintained at

14.degree. C. The contact time of the beer wort in the yeast cake was about 2 minutes.

The apparent degree of fermentation (fermentation-cellar fermentation degree) of the beer emerging from the treatment chamber was 4.2 percent extract, also measured with hydrometer (with due consideration of the temperature correction).

This experiment shows that it is possible by the present invention in a continuous process, with contact times of the beer wort with the yeast being of only about 2 minutes to obtain customary degrees of fermentation which heretofore were obtainable only with contact times of several hours.

CLAIMS:

What is claimed is:

1. An apparatus for accelerated fermentation of liquids comprising a series of chambers having an inlet conduit for introducing a microorganism and an inlet conduit for a liquid to be fermented, each of said chambers being provided with at least one communicating conduit with a next, adjacent chamber, closing means communicating with each chamber for alternatively closing of said inlet conduits, a porous member within each of said chambers upon which the microorganism is deposited, said porous member having pores of a diameter less than the diameter of particle sizes of said microorganism, hydraulic pressure means for forcing said liquid to be fermented through said porous member upon which the microorganism is deposited, a conduit for collecting the fermented liquid communicating with the last chamber, said porous member having a communicating conduit with each chamber wherein the porous member forms the outlet opening of the preceding chamber and the inlet opening for the adjacent chamber.
2. The apparatus according to claim 1, wherein the closing means communicating with each chamber is a slideable pipe conduit having wall perforation, said closing means being disposed in a circular channel conduit communicating with each chamber wherein the wall perforation of the slideable pipe conduit in the open position coincide with the inlet conduit for each of the chambers.
3. The apparatus according to claim 1 wherein the porous body is reinforced with support means.
4. The apparatus according to claims 1, wherein a heat exchange means is provided in each of the chambers for varying the temperature of the treated liquid.
5. The apparatus according to claims 1, wherein inlet means are provided for depositing a different microorganism on each of the porous members.
6. The apparatus according to claim 1 and wherein said porous member has pores of a diameter less than the diameter of a microorganism adsorbed on a carrier therefor.

A further application of the present invention is to the brewing process itself. In FIG. 6, the curve 70 shows s.g. against time and the curve 72 temperature against time during the brewing process (which typically proceeds for a period of 3-5 days). It is conventional to monitor temperature continuously and to apply cooling to the fermenting vessel during the period 72a to limit the maximum temperature. It is also conventional to remove samples from the vessel from time to time to measure s.g. by hydrometer; when a desired s.g. has been reached, the temperature is forced down by additional cooling (period 72b) to terminate fermentation.

Previous attempts to measure s.g. during fermentation by ultrasonic techniques without drawing off samples have been unsuccessful, owing to the presence of gas bubbles and yeast and other solids.

An embodiment of the present invention overcomes these problems by means of a statistical technique. The fermentation period is divided into a number of relatively short time periods, and within each time period a relatively large number of measurements are taken. Suitably the apparatus of FIG. 1 is used and each recorded measurement is the number of pulses passing in a predetermined time interval; such number of pulses is referred to hereinafter as a "count". In the presently preferred embodiment for use with beer, the fermentation is divided into time periods of four minutes, and in each period eight thousand counts are taken.

FIG. 7 illustrates typical statistical distributions for three time periods at different points in the fermentation process. Curve 82 is at an early stage and shows a sharply defined peak. Curve 84 is at a later stage with a high degree to gas generation, and shows a much greater variance. Curve 86 is towards the end of the process where gas evolution has almost ceased, and the variance is again much smaller.

In each case, however, the applicants have established that the distribution contains information which can be used to derive actual s.g. at that time period, and other data. It has been found that the minimum point of the distribution is a function of both s.g. and o.g., and of temperature. Given the relationship between s.g. and o.g. discussed above, and given that o.g. for a particular liquor is known, it is therefore possible to determine s.g. at each time period as fermentation proceeds.

The relationships between these variables can be established empirically by building up a library of data by measuring s.g. by hydrometer for various o.g., temperature and ultrasonic count values. Once such a library is available, it is possible to measure s.g. automatically in an on-line and non-intrusive manner.

The distribution information also provides the possibility of automatically measuring other factors during brewing. As indicated above, the spread of the distribution is affected by outgassing; it is also affected by suspended solids. Similar empirical techniques may be used to quantify these effects, and thus provide on-line data relating to these factors.

It will be apparent that the counts can be stored in digital form and their minima and spread derived in a computer by techniques which are well-known and will therefore not be described here.

CLAIMS:

We claim:

1. A method of determining the specific gravity of a fermentable liquor during fermentation having more than two components and consisting of variable proportions of water, alcohol, dissolved fermentable sugars, yeast solids, and carbon dioxide, the method comprising the steps of fermenting the liquor:

establishing a set of data defining the relationship, at a standard temperature, between constant sonic velocities in fermentable liquors of varying original gravities, measuring the original gravity of a given fermentable liquor before fermentation thereof commences.

fermenting said given fermentable liquor,

measuring the velocity of sound in said given fermentable liquor at a given point in time following commencement of fermentation of said given fermentable liquor,

measuring the temperature of said given fermentable liquor at said point in time, and

deriving from said measured velocity and said measured temperature and said set of data the specific gravity of said given fermentable liquor at said given point in time.

2. A method of determining the specific gravity of a given fermentable liquor having more than two components and consisting of variable proportions of water, alcohol, dissolved fermentable sugars, yeast solids, and carbon dioxide by the second method of claim 1, as fermentation of said given fermentable liquor proceeds, in which:

the time required for said fermentation is divided into a number of time periods each of which is short in relation to the time required for said fermentation, in each said time period a statistically significant number of measurements are made of sonic velocity in said given fermentable liquor, the maximum sonic velocity during each given said time period is established, and the specific gravity of said given fermentable liquor during each given said time period is derived using said maximum velocity as said measured velocity.

3. The method of claim 2, in which the statistical distribution of said statistically significant number of measurements of sonic velocity in said given fermentable liquor in a given said time period is analysed and the spread of said statistical distribution is used as a measure of at least one of solids content and outgassing in said given fermentable liquor in the given said time period.

4. The method of claim 1, in which the given fermentable liquor is contained within a containment defined by a wall, and the sonic velocity in said given fermentable liquor is measured by means of ultrasonic sensors positioned on the exterior of said wall.

5. The method of claim 4, in which said temperature measurement is made by means of a sensor attached to the exterior of said wall and within a thermal insulation.

6. A method of diluting a fermented liquor, said method comprising the steps of:

supplying said fermented liquor to a mixing vessel at a known rate,

supplying water to said mixing vessel at a controlled rate,

mixing said supplied fermented liquor with said supplied water in said mixing vessel to form a product consisting of dilute fermented liquor,

measuring the specific gravity of the product resulting from said mixing, the specific gravity of said product being measured by the method of claim 1, and

controlling the rate of supply of water to said mixing vessel in dependence on the measurement of specific gravity of the product resulting from the said mixing to form a product of controlled dilution.

WEST

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L4: Entry 16 of 37

File: USPT

Sep 25, 1990

US-PAT-NO: 4959228

DOCUMENT-IDENTIFIER: US 4959228 A

TITLE: Measurement of specific gravity during fermentation

DATE-ISSUED: September 25, 1990

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APPL-NO: 7/ 436096

DATE FILED: November 13, 1989

PARENT-CASE:

This application is a continuation of application Ser. No. 070,849 filed 7/31/87; now abandoned.

FOREIGN-APPL-PRIORITY-DATA:

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GB	8526628	October 29, 1985

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FIELD-OF-SEARCH: 426/11, 426/231, 73/32R, 73/32A, 73/452, 137/91

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U.S. PATENT DOCUMENTS

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PAT-NO	ISSUE-DATE	PATENTEE-NAME	US-CL
<input type="checkbox"/> <u>3952761</u>	April 1976	Friedland	73/452
<input type="checkbox"/> <u>4442700</u>	April 1984	Swoboda	73/32A

ART-UNIT: 132

PRIMARY-EXAMINER: Cintins; Marianne

ATTY-AGENT-FIRM: Ratner & Prestia

ABSTRACT:

A method of determining the specific gravity of a fermentable liquor during fermentation is disclosed. Specific gravity is determined from measurements of original gravity, velocity of sound, and temperature.

6 Claims, 7 Drawing figures
Exemplary Claim Number: 1
Number of Drawing Sheets: 5

BRIEF SUMMARY:

This invention relates to a method and apparatus for measuring the specific gravity (s.g.) of liquids. It is of particular utility in relation to brewing, distilling and other processes for producing material for human consumption, but is not limited thereto.

Measurement of s.g. is of great interest in monitoring and controlling brewing and distilling processes. Conventionally, such measurement is done by physically removing a sample for test in a hydrometer or a more sophisticated instrument. It is desirable to replace such measurement by an on-line measurement which could be made part of a process control loop. One object of the present invention is to make this possible.

Accordingly, one aspect of the invention provides a method of measuring the s.g. of a liquid, comprising measuring the velocity of sound in the liquid, measuring the temperature of the liquid, and deriving from these the s.g. of the liquid..

From another aspect, the invention provides apparatus for measuring the s.g. of a liquid, comprising an ultrasonic pulse transmitter and receiver separated by a known path length, means for establishing the transit time of pulses between the transmitter and receiver, a temperature sensor for measuring the temperature of the liquid, and computing means arranged to derive from the transit time and the temperature the s.g. of the liquid.

In preferred forms, the transmitter, receiver and temperature sensor are positioned on the exterior of the liquid container, which may be a storage container such as a tank, or a conduit in which the liquid is flowing.

A relatively simple form of the invention is useful in measuring the s.g. of single-phase solutions such as water/ethanol, for example in whisky distilling. Other embodiments of the invention have utility in measuring the s.g. of more complex solutions and solutions containing gas bubbles and/or solids, for example in brewing.

It will be understood that the apparatus may be used to give a direct reading of a parameter related to s.g., such as the concentration of an acid.

DRAWING DESCRIPTION:

Embodiments of the invention will now be described, by way of example, with reference to the accompanying drawings, in which:

FIG. 1 is a schematic block diagram of one apparatus embodying the invention;

FIG. 2 is a set of graphs representing the relationship between original gravity and specific gravity for beers and the speed of sound therein;

FIG. 3 is a schematic diagram of an apparatus for diluting beer incorporating a monitor embodying the invention;

FIG. 4 is a cross-section of the monitor used in FIG. 3;

FIG. 5 is a graph, similar to FIG. 2, illustrating the operation of the apparatus of FIG. 3, and

FIG. 6 shows graphs of temperature and s.g. against time in a brewing process.

DETAILED DESCRIPTION:

Referring to FIG. 1, a tank 10 holds a liquid 12 the s.g. of which is to be measured. The apparatus includes an ultrasonic transmitting transducer 14 and receiving transducer 16. A pulse generator 18 is connected to drive the transmitting transducer 14 immediately after the preceding pulse has been received at 16.

The arrangement is such that the time taken for a given number of pulses to traverse across the fixed space between the transducers 14 and 16 can be measured from which a measure of the velocity of sound in the liquid 12 can be derived. For this purpose the pulse generator drives a counter 20 to give an output when the predetermined number of pulses has traversed the circuit. The output of the counter 20 is used to reset a counter 22 receiving clock pulses from a clock circuit 24. Thus the output of counter 22 on line 26 is a measure of the time taken for transit of this number of pulses, which is representative of the sonic velocity.

Alternatively, the transmitter and receiver transducers could be located together, or a single transducer used for both purpose, with the pulses being reflected from the opposite wall of the tank.

The apparatus further includes a temperature sensor 28 giving a signal on line 30 representative of the temperature of the liquid 12. The temperature sensor 28 is suitably a thermistor. It has been found that an accurate measure of the temperature of the contents can be made by a thermistor applied to the exterior of the tank provided that the tank wall of sufficiently low thermal conductivity material and is relatively thin (one example being 1/8 inch stainless steel) and the thermistor is enclosed in thermally insulating material, to minimise the effect of ambient air temperature.

The signals on lines 26 and 30 are supplied to a computing means such as a microprocessor 32 which is arranged to compute the specific gravity of the liquid 12. The speed of sound in the liquid is a function of temperature, specific gravity, and compressibility, and since compressibility is for all practical purposes constant for a liquid the s.g. can be computed given knowledge of the sonic velocity and temperature.

The output signal from the microprocessor 32 representing s.g. can be displayed or can be used in a process control loop.

It will be noted that in the drawing the transducers 14 and 16 are shown as being external to the tank 10. This is desirable when one is dealing with substances for human consumption or with substances which are dangerous or corrosive. A technique for permitting such transducers to function via the wall of the tank is fully described in our patent publication No. WO84/01233.

The above embodiment may be modified to minimise the effect of drift and ambient variations in the electronics. As described in our patent publication No. EP 0037196 such modification makes use of a reference path through a medium other than the liquid 12 of interest.

It has been found that the above embodiment provides excellent results in measuring the s.g. of pure liquids and single-phase solutions such as ethanol and water. However, determining the s.g. of multi-phase solutions (e.g. water/ethanol/sugar) and liquids containing gas bubbles or solids presents difficulties.

One example of this is beer. When fermenting beer one starts with a material which is essentially a solution of sugars in water. Fermentation converts all, or more commonly only some, of the sugars to ethanol, and thus during fermentation a three-phase solution is present. Additionally, during fermentation carbon dioxide gas is evolved at a varying rate and yeast solids are present in varying amounts.

The above embodiment has been found not to give useful measurements of actual s.g. in these circumstances.

The applicants have, however, made the unexpected discovery that the above technique can provide an accurate measure of the original gravity of a beer after fermentation. The term "original gravity" (o.g.) denotes the specific gravity of the liquid before fermentation, and is a measure of the amount of sugars present and thus also of the potential alcohol content if these sugars were to be fully fermented to alcohol. A typical beer o.g. would be 1.035, which in the United Kingdom would conventionally be described as "o.g. 1035". A further term used hereinafter is "product gravity" (p.g.) which refers to the specific gravity of the beer after brewing.

Referring to FIG. 2, there is shown in full lines a family of curves showing the s.g. for beers of various o.g. as they progress from unfermented to fully fermented (assuming constant temperature). As one example, the curve 30 represents a beer with an o.g. of 1.035 which, if totally fermented, results in a beer of s.g. 0.97 approximately. If, however, fermentation is stopped at the point 32, a beer of s.g. approximately 1.008 is obtained, some sugars remaining unfermented. The dashed lines in FIG. 2 represent constant sonic velocities (in the absence of gas and solids) and it will be seen that these follow s.g. closely, but not precisely. Thus, if sonic velocity alone is known, it is possible to derive the o.g. of the liquid but not the actual s.g. which may lie anywhere along the curve. The small differences between the constant o.g. and the constant sonic velocity curves may readily be allowed for by building up a library of curves of empirical methods. This data may suitably be set up in computer memory and the computer programmed to make the correction, using interpolation techniques where necessary.

Normally, the o.g. of a brew is known since this is measured by hydrometer before fermentation is started; the apparatus of the present invention can equally be used to measure o.g. Thus, the relationship illustrated in FIG. 2 allows the known o.g. and the measured sonic velocity to be used in principle to derive the s.g. at the time of measurement. Embodiments of the invention making use of this will now be described.

FIG. 3 illustrates an embodiment in which the above factors are utilised to control a dilution or "cutting" process for beer. It is commercially attractive to brew beer to high strength and thereafter dilute it, but it is necessary to control the dilution to give a product equivalent to a stated o.g.

In FIG. 3, beer of a known o.g. in tank 40 is mixed with water from tank 42 in a mixer 44 of known type. The flows are controlled by respective valves 46, 48, the beer control valve 46 normally being set at a constant flow rate and the water control valve 48 being varied as part of a feedback loop. The product passes from the mixer 44 through a flow meter 50 and monitor 52. The monitor 52 acts to measure the temperature and sonic velocity of the product, generate a temperature-compensated sonic velocity, and supply this as a feedback signal to the valve 48. A similar monitor 54 may be provided to check that beer flowing from the tank 40 is of the stated o.g. FIG. 4 shows in greater detail the monitor 52 mounted on a pipe 54. The monitor comprises an ultrasonic transmitter transducer 56, an oppositely-disposed receive transducer 58, and a thermistor 60, all secured to the exterior of the pipe 54 and enclosed in thermal insulation such as an expanded polystyrene sleeve 62. The relationship between the wall thickness and ultrasonic frequency and the mode of use of the ultrasonic transducers is as described in the above-mentioned publications.

FIG. 5, which is a graph similar to FIG. 2, illustrates the operation of the apparatus of FIGS. 3 and 4. The beer in tank 40 has a known o.g. and a known s.g. (measured, for example, by hydrometer) and is thus defined by point 64. Dilution with water causes the characteristics of the product to move along the chain-dotted line 66. By maintaining the temperature-corrected ultrasonic count within the curves 68 and 70, a product is achieved which is equivalent to a beer of o.g. at area 72 and alcohol content at area 74.

It will be appreciated that FIG. 3 shows only the apparatus used when operating as a continuous process. For starting up the process, additional apparatus (not shown) may be required, such as means for feeding product back to the mixer until a relatively stable control feedback is achieved.